

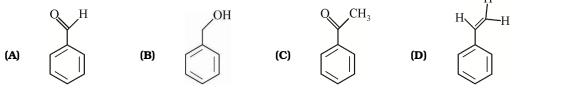
Date Planned ://	Daily Tutorial Sheet-7	Expected Duration : 45 Min
Actual Date of Attempt ://	JEE Advanced (Archive)	Exact Duration :

PARAGRAPH FOR QUESTIONS 91 - 92

In the following reaction sequence, the compound ${\bf J}$ is an intermediate. ${\bf I} \xrightarrow{{\rm (CH_3CO)_2\,O}\atop{\rm CH_3COONa}} {\bf J} \xrightarrow{{\rm (i)}\,{\rm H_2,Pd/C}\atop{\rm (ii)}\,{\rm SOCl_2}\atop{\rm (iii)}\,{\rm anhyd.\,AlCl_3}} {\bf k}$

 $\mathbf{J}(C_9H_8O_2)$ gives effervescence on treatment with NaHCO3 and a positive Baeyer's test.

89. The compound **I** is : (2012)



90. The compound **K** is: (2012)

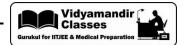
*91. Among P, Q, R and S, the aromatic compound(s) is(are):

92. The major product H in the given reaction sequence is : (2012)

93. The number of aldol reaction(s) that occurs in the given transformation is: (2012)

(A)

(A)



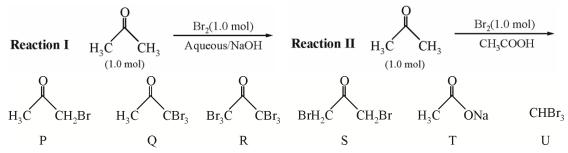
94. The product of acid hydrolysis of P and Q can be distinguished by :

(2013)

- (A) Lucas reagent
- **(B)** 2, 4-DNP
- (C) Fehling's solution
- (**D**) NaHSO₃

- $P: H_2C \longrightarrow \begin{pmatrix} OCOCH_3 \\ CH_3 \end{pmatrix}; \qquad Q: H_3C \longrightarrow OCOCH$
- 95. After completion of the reactions (I and II), the origin compound(s) in the reaction mixtures is(are):

(2013)



- (A) Reaction I: P and reaction II: P
- (B) Reaction I: U, acetone and reaction II: Q, acetone
- (C) Reaction I: T, U, acetone and reaction II: P
- **(D)** Reaction I: R, acetone and reaction II: S, acetone
- **96.** The major product in the following reaction is :

(2014)

C1
$$CH_3 \xrightarrow{1. CH_3 MgBr, dry ether, 0^{\circ}C} C$$

$$(A) \qquad H_3C$$

$$CH_3 \qquad (B) \qquad H_2C$$

$$CH_3 \qquad (CH_3) \qquad (CH_3)$$

$$CH_3 \qquad (CH_3) \qquad (CH_3)$$

PARAGRAPH FOR QUESTIONS 97 - 98

Schemes 1 and 2 describe sequential transformation of alkynes M and N. Consider only the major products formed in each step for both the schemes.

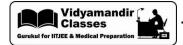
HO

H

$$\begin{array}{c}
1. \text{ NaNH}_2 \text{ (excess)} \\
2. \text{ CH}_3\text{CH}_2\text{I (1 equivalent)} \\
3. \text{ CH}_3\text{I (1 equivalent)} \\
4. \text{ H}_2, \text{ Lindlar's catalyst}
\end{array}$$

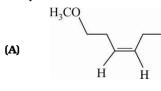
Scheme-1

$$\begin{array}{c}
1. \text{ NaNH}_2 \text{ (2 equivalent)} \\
\hline
0. \text{ Particle of the problem of th$$



97. The product X is :

(2014)



(B) H₃CO

(C) CH₃CH₂O

(**D**) CH₃CH₂O

98. The correct statement with respect to product Y is :

(2014)

- (A) It gives a positive Tollens test and is a functional isomer of X
- (B) It gives a positive Tollens test and is a geometrical isomer of X
- (C) It gives a positive iodoform test and is a functional isomer of X
- (D) It gives a positive iodoform test and is a geometrical isomer of X
- **99.** Consider all possible isomeric ketones including stereoisomers of MW = 100. All these isomers are independently reacted with $NaBH_4$. The total number of ketones that gives a racemic product(s) is(are)_____. (2014)
- **100.** The major product of the following reaction is :

(2015)

*101. Among the following, the number of reaction(s) that produce(s) benzaldehyde is ______. (2015)

(D)

- (A) CO, HCl
 Anhydrous AlCl₃/CuCl
- (B) $H_2O \longrightarrow 100^{\circ}C$

(C) $\begin{array}{c} & & \\ & & \\ \hline \\ & & \\ \hline \\ Pd-BaSO_4 \end{array}$

- COOMe

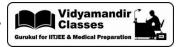
 DIBAL-H

 Toluene, -78°C

 H₂O
- **102.** The number of hydroxyl group(s) in Q is:

(2015)

$$\begin{array}{ccc} & & & & & & \\ & & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ &$$



 $\textbf{103.} \quad \text{Consider the reaction sequence below:} \\$

(2016)

$$\underbrace{\text{Succinic anhydride}}_{\text{AlCl}_3} \land \underbrace{\text{Clemmensen's}}_{\text{reduction}} \lor X$$

X is:

(C)
$$H_3CO$$